Atomic Positions in U₃Si

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The positions of the atoms in the unit cell of U₃Si have been re-examined. The best fit with X-ray diffraction spectra was obtained in the space group $I4/mcm(D_{4h}^{18})$ [origins (0,0,0), $(\frac{1}{2},\frac{1}{2},\frac{1}{2})$] with the following atomic positions: four U(1) at $(0, \frac{1}{2}, \frac{1}{4})$, $(\frac{1}{2}, 0, \frac{1}{4})$; eight U(2) at $\pm (u, u, +\frac{1}{2}, 0)$, $\pm (u + \frac{1}{2}, \bar{u}, 0)$ $(u = 0.226 \pm 10.01)$ 0.002); four Si at $\pm (0, 0, \frac{1}{2})$.

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Introduction

One of the few fuel materials that can be used without enrichment is the δ phase of U₃Si. Zachariasen (1949) was the first to identify the space group and the crystal structure of U₃Si. He determined lattice size and atom positions by the X-ray diffraction (Debye-Scherrer) technique. Later, other investigators (Blum, Silvestre & Vaugoveau, 1965; Boucher, 1971; De Vooght, Verniers & De Meester, 1973) re-examined the tetragonal lattice parameters. In recent work the

lattice parameters were determined very carefully (Boucher, 1971; De Vooght et al., 1973) but the observed X-ray intensities showed unexplained deviations when compared with the calculated intensities based on the suggestion of Zachariasen for the atomic positions. The present study deals with the replacement of the various atoms in the lattice in order to achieve better agreement between the observed and the calculated X-ray diffraction intensities.

Table 1. Comparison between observed and calculated intensities assuming atomic positions according to Zachariasen (1949) after refinement of U(2) atomic positions

Fig. 1. Metallography of U_3Si (× 500).

h	k	1	Iobs	Ical	Fobs	Fcat
0	0	2	10.5	8.346	306.022	287.800
1	1	0	17.5	14.378	284.781	272.296
1	1	2	13.2	15.649	250.805	288.066
2	0	0	5.7	5.547	234.373	244.976
2	1	1	8.9	8.320	175.677	178.677
2	0	2	100.0	100.000	868.011	915-651
0	0	4	19.3	18.220	887.467	909.594
2	2	0	29.8	30.735	796.728	853.541
2	1	3	4.3	3.864	166-423	165.838
1	1	4	6.8	3.628	301.964	232.684
2	2	2	2.9	2.486	199.826	195.176
3	1	0	3.7	5.105	226.697	280.910
2	0	4	3.7	2.433	249.193	213.456
3	1	2	8.9	4.247	277.514	201.663
3	2	1	1.9	1.800	137.143	141-540
2	2	4	24.2	21.349	764.790	757.755
4	0	0	10.0	9.158	703.086	709.763
2	1	5	2.8	1.485	191.666	148-576
0	0	6	0.4	0.403	211.570	221.171
3	2	3	1.1	1.196	123.020	134.692
4	1	1	6.4	5.199	296.885	282.273
3	1	4	3.4	4.118	218-109	253-214
4	0	2	0.9	0.546	158.080	131-355
4	2	0	0.2	0.338	150.445	110.276
3	3	2	0.9	2.655	168.328	308-450
1	1	6	1.2	1.494	193-213	227.380
3	3	0	1.6	0.242	302.178	123-944
4	1	3	6.0	3.774	323-179	270.387
2	0	6	16.0	13.767	746-613	730.560
4	2	2	18.6	20.864	577.687	645-416
4	0	4	9.1	9.779	596-917	651-308



Experimental

A nominally uranium 3.8 wt.% silicon alloy was melted, in a vacuum of 10⁻⁵ torr, in zirconia crucibles and cast into cold copper moulds. The chemical analysis showed the impurities: Zr (300), Fe (50), Cr (50), Ni (50), Cu (20), Mn (20 ppm). The weight percent of the silicon was 3.85 ± 0.05 %. The ingot (φ 22 × 50 mm) was annealed *in vacuo* for several days to obtain U₃Si by the peritectoid reaction: $U_3Si_2 + 3U$ $\rightarrow 2U_3Si$. The annealed billets where cut (in various directions) into disc-shaped samples of 22 mm diameter and 3 mm thickness. The flat side of these samples was ground and then polished electrolytically to remove any possible damage layer on the surface. The microstructure, as observed by X-rays and by metallography, was randomly oriented polycrystalline (Fig. 1). The grains had substructure, as a result of the martensitic transformation during cooling. This martensitic substructure does not affect the X-ray diffraction patterns (De Vooght et al., 1973).

A vertical Philips X-ray goniometer was used. The recording (Cu K α radiation) was continuous. The goniometer angular (2 θ) velocity was $\frac{1}{8}$ deg min⁻¹, in order to obtain reliable values of integrated X-ray intensities. Divergence and receiving slits were 1°. The detector was a proportional counter.

Table 2. Comparison between observed and calculated intensities assuming atomic positions according to the present work, after refinement of U(2) atomic positions

h	k	1	I_{obs}	I_{ca1}	F_{obs}	F_{cal}
0	0	2	10.5	8.341	304.415	287.800
1	1	Ō	17.5	17.689	283.286	302.099
1	1	2	13.2	12.715	249-488	259.723
2	0	0	5.7	5.555	233.142	245·211
2	1	1	8.9	8.250	174.755	177.971
2	0	2	100.0	100.000	863.454	915.876
0	0	4	19-3	18.211	882.808	909.594
2	2	0	29.8	30.750	79 2 ·545	853-953
2	1	3	4.3	3.832	165.549	165.185
1	1	4	6.8	4.456	300.378	2 57·908
2	2	2	2.9	2.495	198.776	195.574
3	1	0	3.7	2.794	225.506	207 .856
2	0	4	3.7	2.437	247.884	213.659
3	1	2	8.9	7.753	276.057	272.540
3	2	1	1.9	1.787	136.423	141.098
2	2	4	24.2	21.359	760.774	758.121
4	0	0	10.0	9.172	699.395	710-492
2	1	5	2.8	1.473	190.660	147.988
0	0	6	0.4	0.402	210.459	221 .171
3	2	3	1.1	1.188	122.374	134·270
4	1	1	6.4	5.160	295.327	281·271
3	1	4	3.4	2.258	2 16·964	187.556
4	0	2	0.9	0.552	157-250	1 32·0 67
4	2	0	0.7	0.343	149.655	111.091
3	3	2	0.9	0.416	167.445	122.129
1	1	6	1.2	1.216	19 2 ·198	205.238
3	3	0	1.6	1.555	300.591	314.374
4	1	3	6.0	3.746	321.483	269.423
2	0	6	16· 0	13.767	742.693	730.739
4	2	2	18.6	20.905	574.654	646-215
4	0	4	9.1	9.794	593.783	651.977

Results and discussion

The X-ray intensities of the *hkl* reflexions are given in Table 1. For the calculation of the structure factors, the uranium and the silicon were assumed to be ionized [4 + and 2 +, respectively (Zachariasen, 1949)]. Atomic structure factors were taken from the data of Hanson, Herman, Lea & Skillman (1964). For the temperature factor recent results for the Debye temperature for U₃Si were used (Rosen, Gefen, Kimmel & Halwany, 1973). The lattice parameters are a = 6.033, c = 8.690 Å (De Vooght *et al.*, 1973). The atomic positions of U(2) were refined by finding the *u* value for the smallest residual R ($R = \sum |F_{cal} - F_{obs}| / \sum F_{obs}$). In the first stage the positions of U(1) and Si were

In the first stage the positions of U(1) and Si were placed according to Zachariasen (1949) and then the U(2) positions were refined. Table 1 shows the improved agreement between the calculated and the observed spectra when u is 0.226 (compared with 0.231, as proposed by Zachariasen). Most of the typical dis-



Fig. 2. The residual versus U(2) atomic positions. Upper curve: previous arrangement (Zachariasen, 1949). Lower curve: new arrangement (present).



Fig. 3. U₃Si structure viewed along the fourfold axis. Large circles: Si atoms. Small filled circles: U(1) atoms. Small open circles: U(2) atoms.



Fig. 4. Face-centred tetragonal U_3Si sublattice. Large circles: Si atoms. Small filled circles: U(1) atoms. Small open circles: U(2) atoms.

crepancies pointed out by De Vooght *et al.* (1973) appear in the 114, 310, 312, 330, 332 reflexions.

In order to obtain better agreement between the observed intensities and the calculated spectra, we tried to interchange the positions of the U(1) atoms with those of the silicon, and then refine the positions of the U(2) atoms. Table 2 shows that in this case the observed intensities are in good agreement with the calculated values. Most of the anomalies reported by De Vooght disappeared; for example, there was a good fit for the intensities of 211, 213, 312, 321, 330; fair agreement was found for 215, 411, 310. The average residual R improved from 10.6 to 6.9 % (Fig. 2). The parameter u was determined as 0.226 ± 0.002 . The new atomic positioning is therefore (Fig. 3), for the origins (0,0,0), $(\frac{1}{2},\frac{1}{2},\frac{1}{2})$; four U(1) at $(0,\frac{1}{2},\frac{1}{4})$, $(\frac{1}{2},0,\frac{1}{4})$; eight U(2) at $\pm (u,u+\frac{1}{2},0)$, $\pm (u+\frac{1}{2},\overline{u},0)$; four Si at $\pm (0,0,\frac{1}{4})$. This new determination of the atomic positions gives the new interatomic distances shown in Table 3.

The new arrangement shows much less variation in the distances between the uranium and the neighbouring silicon atoms. The largest, as well as the shortest distances are between the U(1) and U(2) atoms. The metallic radii of the silicon and the uranium in the U₃Si are equal $(1.53 \pm 0.05 \text{ Å})$. The valence of the U and Si ions is approximately the same as that given by Zachariasen (1949). The f.c.t. subcell in the U₃Si unit cell is considered to be a distorted Cu₃Au type, where the U(2) atoms are not on the {100} plane, but a little outside or inside the {100} faces, as shown in Fig. 4.

Table 3. Interatomic distances

The previous values (Zachariasen, 1949) are given in parentheses.

U(1)-4 Si	3.016 Å	(3·01 Å)
-4 U(2)	2.93	(3.04)
-4 U(2)	3.17	(3.04)
U(2)-4 Si	3.05	(2.92, 3.17)
-2 U(1)	2.93	(3.04)
-2 U(1)	3.17	(3.04)
-4 U(2)	3.05	(3.01)
Si4 U(1)	3·0 16	(3.01)
-8 U(2)	3.05	(2.92, 3.17)

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